CHEMICAL AND THERMAL STABILITY OF REFRIGERANT-LUBRICANT MIXTURES WITH METALS

Final Report

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TABLE OF CONTENTS

ABSTRACT	1
SCOPE	2
SIGNIFICANT RESULTS	3
INTRODUCTION	3
DISCUSSION OF RESULTS HFC/PE Ester Mixtures HFC/PAG Mixtures HCFC/Hydrocarbon Lubricant Mixtures	5 5 7 9
CONCLUSIONS	11
TEST PROCEDURES	14
SEALED TUBE TESTS	14
EVALUATION OF AGING TESTS Visual Inspection Gas Chromatographic (GC) Analysis Chloride and Fluoride Ion Determination Oil Analysis	14 14 15 16
COMPLIANCE WITH AGREEMENT	18
PRINCIPAL INVESTIGATOR EFFORT	18
APPENDICES	

A - Refrigerant/Lubricant Test Data
B - Oil Analysis Data
C - Refrigerant and Lubricant Lists

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ABSTRACT

This report presents the results of a sealed tube stability study on twenty-one refrigerant-lubricant mixtures selected from the following groupings:

- HFCs R-32, R-125, R-134, R-134a, R-143a, and R-152a with one or more lubricants selected from among three pentaerythritol esters and three polyalkylene glycols. All lubricants were carefully pre-dried to 25 ppm or less moisture content.

- HCFCs R-22, R-123, R-124, and R-142b, as well as CFC R-11, with one or more lubricants selected from among two mineral oils and one alkylbenzene fluid.

Each test mixture was aged at three temperature levels.

Primary results obtained were:

1. HFCs tested, as well as R-22, are very stable and do not undergo chemical or thermal decompositions after 14 days at temperatures up to 200°C (392°F).

2. HCFCs R-124 and R-142b have stability properties intermediate between above HFCs and the currently widely used CFC R-12.

3. HCFC R-123 while more reactive than any of the above, still offers a stability improvement by a factor of ten over R-11, the CFC refrigerant it is to replace in low pressure chiller applications.

4. All pentaerythritol ester and all polyalkylene glycol fluids showed some changes of chemical structure after exposure to 200°C (392°F), and in some cases after exposure to the lower temperatures of 150°C (300°F) and 175°C (347°F).

This report supersedes previously published quarterly reports DE/CE/23810-2A and DE/CE/23810-3A.

The scope of ARTI MCLR Project 650-50200 encompasses the determination of the chemical and thermal stability of twenty-one refrigerant-lubricant combinations. The test methodology chosen was the sealed tube stability test. Refrigerants and lubricants are heated at elevated temperatures in the presence of a steel catalyst. Test results are based on chemical and physical measurements designed to assess the changes that occur to the test materials during thermal aging.

Refrigerants tested as part of the project consisted of six HFCs, four HCFCs, and one CFC; the latter was only included to establish base line information. Lubricants tested were three polyglycols, three polyolesters, two mineral oils and one alkylbenzene fluid.

The goal of the project was an assessment of the general stability properties of generically selected refrigerant-lubricant mixtures, rather than endorsement (or opposite thereof) of any of the specific materials tested. Aging tests were conducted over temperatures ranging from 105°C (220°F) to 200°C (392°F) depending on the specific materials combinations tested. As it will be apparent from the results, not all materials tested perform equally well at the higher temperatures. On the other hand, not all refrigerant systems require materials capable of 200°C performance. Chemical and thermal stability of refrigerant-oil mixtures is but one of a number of properties necessary for reliable performance in refrigeration and air-conditioning systems.

SCOPE

SIGNIFICANT RESULTS

INTRODUCTION

A total of twenty-one refrigerant-mixtures were tested during this study. They were:

- R-22/Mineral Oil-ISO VG32 (150 SSU viscosity)
- R-32/Pentaerythritol Ester (mixed acid I)
- R-32/Polypropylene Glycol (butyl monoether)
- R-123/Mineral Oil-ISO VG32 (150 SSU viscosity)
- R-124/Alkylbenzene ISO VG32 (150 SSU viscosity)
- R-125/Polypropylene Glycol (butyl monoether)
- R-125/Modified Polyglycol
- R-125/Pentaerythritol Ester (mixed acid I)
- R-134a/Polypropylene Glycol (butyl monoether)
- R-134a/Polypropylene Glycol Diol
- R-134a/Modified Polyglycol
- R-134a/Pentaerythritol Ester (mixed acid II)
- R-134a/Pentaerythritol Ester (mixed acid I)
- R-134a/Pentaerythritol Ester (100 cSt)
- R-142b/Alkylbenzene ISO VG32 (150 SSU viscosity)
- R-143a/Pentaerythritol Ester (mixed acid I)
- R-152a/Alkylbenzene ISO VG32 (150 SSU viscosity)
- R-134/Pentaerythritol Ester (mixed acid I)
- R-11/Mineral Oil-ISO VG32 (150 SSU viscosity)
- R-11/White Naphthenic Mineral Oil (ISO VG45)
- R-123/White Naphthenic Mineral Oil (ISO VG45)

Test results are presented in the form of individual data tables for each refrigerant-lubricant mixture. These tables are consolidated in Appendix A (Tables A-1 through A-21). Following each data table are the gas chromatograms for the test samples covered in each table (Figures A-1 to A-42).

Listed in the data tables is the following information:

- contents of the sealed tubes reported in each table
- aging conditions

- visual observations on sealed tube contents after aging, including change of oil (liquid phase) color expressed in units of the ASTM Standard D1500 color scale, condition of the steel catalyst, formation of insoluble materials, etc.

- chemical analysis results on aged tubes, including gas chromatography (GC) of the vapor phase tube contents, chloride

or fluoride ion content, and oil analysis results for those mixtures containing polyglycol or polyol ester lubricants: total acid number (TAN), infrared (IR) spectroscopy, and size exclusion chromatography (SEC).

Under "Chemical Analysis" the column following chloride/ fluoride ion results is labeled "% Refrigerant Reacted". This value was calculated from the measured halide ion content by making the assumption that each halide ion generated equals the destruction of one refrigerant molecule. Based on past experience, this is a reasonable assumption at least at low and intermediate levels of reactivity. It is not valid for very reactive systems where more than one halide atom has been removed from refrigerant molecules.

Figures A-1 to A-42 are copies of the chromatograms obtained on the vapor content of aged sealed tubes. There are four chromatograms for each refrigerant-lubricant mixture: unused refrigerant and vapor samples aged at the three test temperatures.

Oil analysis of aged synthetic lubricants consisted of IR spectroscopy for the polyol ester samples, and SEC for the polyalkylene glycol samples. Copies of IR spectra and SEC chromatograms can be found in Appendix B (Figures B-1 to B-22).

Aged sealed tubes and their individual steel catalysts (both aged and unaged) were photographed for all test sets. Copies of photographs from those test sets that exhibited significant visual changes after aging are shown in Appendix D (Figures D-1 to D-8). Appendix D is included only with the permanently bound copies of this report.

All test materials used in this study are referred to by generic designations only. Cross references from generic to commercial designations for all lubricants tested are shown in Appendix C. Also included in Appendix C are the results of purity measurements obtained on all of the refrigerants and lubricants, as received by us. This includes GC analysis of refrigerants, and water, copper and iron content, as well as the total acid number of lubricants.

DISCUSSION OF RESULTS

Discussion of results is arranged in the order HFC/ pentaerythritol (PE) ester mixtures, HFC/polyalkylene glycol (PAG) mixtures, and HCFCs.

HFC/PE Ester Mixtures

Table 1, page 6, is a summary of results taken from the individual data tables for the eight mixtures listed. The R-152a/alkybenzene mixture has been included in this ester result table to avoid an extra classification for just one HFC/alkylbenzene mixture.

The data listed in Table 1 do not show any evidence of significant refrigerant decomposition. This holds for all test mixtures and at all aging conditions. Fluoride contents are very low and very close to the "zero" value for the test method. Values of "% Refrigerant Decomposition" for the titration blanks performed during the analyses are in the range of 0.001 to 0.002%. The actual fluoride concentrations measured (after subtraction of the titration blank) are of the same low order of magnitude.

GC analyses (Appendix A) for all Table 1 mixtures lend additional support to the above findings. Again, there is no evidence of any measurable refrigerant decomposition for any of the test mixtures.

The only changes of any significance observed with the HFC-ester mixtures are on the aged lubricants. Acid numbers of aged oils increased after all of the 200°C (392°F) tests. The increases range from 0.49 to 1.69 units (ASTM Method D664 "Test of Neutralization Number by Potentiometric Titration"). Some of the mixtures also had acid number increases at 150°C (300°F) and at 175°C (347°F). The consistent acid number increases at the highest temperature are believed to be indicative of the thermal stability limits of the fluids tested.

Reasons for the randomly occurring changes at lower temperatures are not quite apparent, except in the case of the R-134a/100 cSt ester mixture. Here, in addition to the acid number changes, IR spectral changes occurred at all test temperatures. The observed IR changes are typical of C=O absorption (formation of free acids) and C-O absorption (free alcohol functions). In addition, GC analysis of 200°C tubes (Figure A-28) detected the presence of CO₂ which is a product of the decarboxylation of ester linkages.

TABLE	1:	STABILITY	RESULTS	HFC/PE	ESTER	LUBRICANTS
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TEST SAMPLE	DATA TABLE	AGING TEMP. (°C)	% REFRIGERANT REACTED (calc. from F ion data)	LUBRICANT ACID NUMBER (mg KOH/g)
R-32/PE (mixed acid)	A-2 1 1 2	50 (300°F) 75 (347°F) 00 (392°F)	0.002 0.001 0.001	0.00 0.00 1.67
R-125/PE (mixed acid I)	A-8	150 175 200	0.005 0.005 0.005	0.00 0.48 1.09
R-134a/PE (mixed acid 11)	A-12	150 175 200	0.000 0.000 0.000	0.00 0.00 0.67
R-134a/PE (mixed acid I)	A-13	150 175 200	0.001 0.001 0.001	0.00 0.00 0.74
R-134a/PE (100 cSt)	A-14	150 175 200	0.001 0.001 0.004	0.94 0.63 0.99
Note: Aged lu spectrum changes a	bricant f fter agin	from this m g at all te	ixture showed sigr emperatures.	nificant IR
R-143a/PE (mixed acid I)	A-16	150 175 200	0.001 0.002 0.001	0.51 1.07 0.99
R-134/PE (mixed acid I)	A-18	150 175 200	0.001 0.000 0.000	0.59 0.83 1.69
R-152a/ALKYL- -BENZENE	A-17	150 175 200	0.000 0.001 0.000	Not applicable Not applicable Not applicable

The only other IR spectral changes observed on polyol esters were minor changes after aging at 200°C (392°F) with the R-125/PE (mixed acid I) mixture.

The overall conclusions from the HFC/PE ester tests are as follows:

1. All HFCs tested are stable at the test conditions used.

2. The high viscosity ester (100 cSt), tested with R-134a, exhibited signs of molecular changes at all three test temperatures. It is unknown at this point whether these observed changes will impact on full scale equipment performance.

3. The other two PE esters tested, mixed acid I and mixed acid II, showed little, if any, changes except for consistent increases in acid numbers after aging at 200°C (392°F). In other words, these two esters appear to be quite stable to at least 175°C (347°F).

4. All tests were performed with esters having very low moisture levels in the 25 ppm range. Results reported here may or may not be representative of the stability of PE esters in the "real world", where much higher water contents are expected to be the norm.

5. No observable reactions occurred with the R-152a/ alkylbenzene mixture.

HFC/PAG Mixtures

Table 2, page 8, summarizes the test results from mixtures containing PAG lubricants. Again, there is no evidence of HFC decomposition by either GC or fluoride ion analysis under any of the three test conditions.

However, size exclusion chromatography (SEC) suggests changes occurred to the structure of the PAG molecules. This was observed to varying degrees with all three PAG lubricants tested, and at all test temperatures. As was noted under the polyol esters, the significance of these observed changes on the actual performance of the PAG lubricants is not clear at the present time. The absence of HFC decomposition products suggests that PAG molecular changes do not give rise to refrigerant instability. The most likely effects, if any, of the PAG changes would be viscosity decreases, which impact lubrication performance, and/or the formation of high molecular weight solids and sludges, which can cause systems flow restrictions.

TABLE 2: STABILITY RESULTS HFC-PAG LUBRICANTS

TEST SAMPLE	DATA TABLE	AGING TEMP. (°C)	<pre>% REFRIGERANT REACTED (calc. from F data)</pre>	Size Exclusion Chromatography
R-32/PPG BUTYL MONOETHER	A-3	150 (300°F) 175 (347°F) 200 (392°F)	0.000 0.002 0.004	Minor changes in PAG structure (all tests)
R-125/PPG BUTYL MONOETHER	A-6	150 175 200	0.001 0.002 0.002	Changes in PAG structure (all tests)
R-125/MODIFIED POLYGLYCOL	A-7	150 175 200	0.007 0.004 0.005	Changes in PAG structure (all tests)
R-134a/PPG BUTYL MONOETHER	A-9	150 175 200	0.001 0.001 0.001	Changes in PAG structure (all tests)
R-134a/PPG DIOL	A-10	150 175 200	0.000 0.001 0.000	Significant changes in PAG structure (all tests)
R-134a/MODIFIED POLYGLYCOL	A-11	150 175 200	0.001 0.001 0.001	Significant changes in PAG structure (all tests)

The test results on HFC-PAG mixtures lead to the following conclusions:

 All HFCs tested are stable at the test conditions used.
 Based on SEC analysis all PAGs tested exhibited varying degrees of molecular change. Again, the impact, if any, of these observed changes on full scale systems performance is unknown.

3. Also unknown at this point are the effects of moisture contents higher than the low 25 ppm level present in the test work reported here.

HCFC/Hydrocarbon Lubricant Mixtures

Table 3, page 10, summarizes the test results from the HCFC containing mixtures. It is apparent that, with the possible exception of R-22, the HCFCs tested are more reactive under the test conditions used than any of the previously discussed HFCs. Both chloride ion and vapor phase GC analyses support this contention in all cases. However, in comparison to CFCs, such as R-12, both R-124 and R-142b exhibit improved chemical and thermal stability properties. Hence, the use of these HCFCs in full scale systems should not present chemical liability questions beyond those that the industry has accepted historically with R-12 and other CFCs.

The fourth HCFC tested, R-123, is significantly more reactive than are any of the "new" refrigerants included in this project, especially at temperatures above 150°C. It must be noted, however, that R-123 is a low pressure refrigerant and will be used as a replacement for R-11 in low pressure systems. There, it will never be exposed to the higher range temperatures of the sealed tube tests.

The GC results obtained on the vapor phase contents of aged R-123 sealed tubes show the expected decomposition path for this refrigerant, namely

CF_3CHCl_2	>	CF ₃ CH ₂ Cl>	CF_3CH_3
R-123		R-133a	R-143a

Unfortunately, pure samples of R-133a were not available for a quantitative calibration of GC results. All R-133a values listed in the tables are uncorrected readouts and are approximate values only.

TEST SAMPLE	DATA TABLE	AGING TEMP. (°C)	<pre>% REFRIGERANT REACTED (calc. from C1 data)</pre>	GC RESULTS
R-22/Mineral Oil (ISO VG32- 150 SSU)	A-1	150 (300°F) 175 (347°F) 200 (392°F)	0.001 0.003 0.004	No changes No changes No changes
R-123/Mineral Oil (ISO VG32- 150 SSU)	A-4	105 (220°F) 150 160 (320°F) 175	0.045/0.032 1.56/2.71 26.0 40.5	No changes R-133a: 1.8/3.4% R-133a: 14.0% R-143a: 0.45% R-133a: 44.5% R-143a: 2.0%
R-124/Alkyl- benzene (ISO VG32)	A-5	150 175 200	0.008 0.014 0.052	No changes No changes No changes
R-142b/Alkyl- benzene (ISO VG32)	A-15	150 175 200	0.011 0.015 0.169	No changes No changes No measurable changes

TABLE 3: STABILITY RESULTS HCFC-HYDROCARBON LUBRICANTS

For Comparison with Past Practices:

R-12/Mineral Oil	175	0.2 (typical	R-22 value) ⁽¹⁾
R-12/Alkylbenzene	175	0.1 (typical	R-22 value) ⁽¹⁾

(1) "Sealed Tube Stability Test Results: Alternative Refrigerants"
 Carrier Corporation, Syracuse, NY, September 1989

Because of the relatively limited high temperature capability of R-123, a direct comparison was established for this refrigerant with the R-11 low pressure CFC refrigerant. Table 4, page 12, summarizes the results obtained from the two refrigerants after aging at 105, 150, and 160°C (220, 300, and 320° F).

The data show that R-123 is approximately ten times more stable than is R-11. Therefore, the use of R-123 as an R-11 replacement is expected to cause fewer, if any, chemical and/or thermal stability problems than does R-11 now in equivalent applications. The reactivity of both refrigerants depends on oil quality, especially at the lower sealed tube temperatures. At 150°C (300°F) and above most, if not all, R-11 was destroyed after 14 days. In addition, all chlorine was removed from some of the R-11 molecules resulting in CO_2 formation (GC analysis). R-123 was not subject to such destructive action under the same test conditions. As a rough comparison, R-123 exhibits chemical activity at 105°C (220°F) similar to that of R-12 at 175°C (347°F).

The HCFC tests lead to the following conclusions:

1. R-22 has excellent stability, at least up to 200°C (392°F), similar to the HFCs tested under this program.

2. R-124 and R-142b are slightly more reactive than the HFCs, but are less reactive than R-12 and similar CFCs.

3. While R-123 was the most reactive new refrigerant tested during this program, it is still significantly less reactive than is R-11 - by a factor of approximately ten. Therefore, the replacement of R-11 by R-123 in low pressure refrigeration equipment will cause fewer, if any, chemical problems than may exist now.

CONCLUSIONS

The following summary conclusions can be drawn from the results reported here:

1. All HFCs tested, as well as R-22, are very stable materials and do not undergo significant chemical reactions or thermal decomposition at temperatures up to 200°C (392°F).

TABLE 4: STABILITY COMPARISON R-123 VS. R-11

TEST SAMPLE	DATA TABLE	AGING TEMP. (°C)	%,REFRIGERANT REACTED (calc from C1 data)	GC RESULTS
R-123/Mineral Oil (ISO VG32)	A-4	105 (220°F) 150 (300°F) 160 (320°F)	0.045/0.032 1.56/2.71 26.0	No changes R-133a: 1.8/3.4% R-133a: 14.0%, R-143a: 0.45%
R-11/Mineral Oil (ISO VG32)	A-19	105	0.46	No measurable change
		150	72.0	R-21: 72%, R-31: <0.5%
		160	87.0	R-11: 0%, R-21: 47.8% R-31: 17.2% CO ₂ : 31.4%
R-11/White Naphthenic Oil (ISO VG46)	A-20	105 150	0.014 45.2	No changes R-21: 17.8%, R-31: 1.37% CO2
		160	97.0	Trace R-11, R-21: 75.0%, R-31: 10.4%, CO ₂ :13.7%
R-123/White Naphthenic Oil (ISO VG46)	A-21	105 150	0.002 5.84	No changes R-133a: 3.9% R-143a: 0.08%
(100 0010)		160	8.85	R-133a: 10.9% R-143a: 0.38% CO ₂

2. Two HCFCs tested, R-124 and R-142b, were found to be intermediate in reactivity between the above HFCs and CFCs, such as R-12.

3. While R-123 is significantly more reactive than is any of the "new" refrigerants tested here, it still offers a stability improvement by a factor of ten over R-11, the CFC refrigerant it is intended to replace in low pressure chiller applications.

4. The two low viscosity (ISO VG22 and VG32) PE esters included in the program exhibited some increases in acid number, especially after aging at 200°C (392°F). The high viscosity ester (ISO 100) tested exhibited additional evidence of molecular changes during aging. Formation of CO_2 at 200°C (392°F) indicates decarboxylation of the ester at that temperature.

5. All polyalkylene glycol lubricants exhibited some signs of molecular change after aging. The significance of these changes has not been established.

TEST PROCEDURES

SEALED TUBE TESTS

All sealed tube tests were conducted in heavy walled glass tubes, 6.35 mm I.D. and 20 cm long $(\frac{1}{4}" \ge 8")$ with one end sealed. After thorough cleaning with toluene and drying at 150°C (300°F) the tubes were necked 5 cm (2") from the open end. This gives finished sealed tubes with an internal free volume of 7.5-8.0 ml (about 0.5 in³).

Metal coupons were cut from valve steel (Sandvik Steel Company), 6.3 mm (¼") wide and 32 mm (1.25") long. After cleaning with hexane the coupons were stored submerged in this solvent. One coupon was added to each tube immediately prior to necking. After necking of the tubes 1.0 ml of lubricant was added using a calibrated syringe. Refrigerant equal to the lubricant weight was next condensed from a gas handling manifold into the liquid nitrogen cooled tubes. After sealing and flame annealing of the seals the finished tubes were inspected for any visible faults. Acceptable tubes were placed into individual slots in large aluminum "blocks" of 12.7 cm (0.5") OD.

Three sealed tubes were prepared for each test set, that is, for every refrigerant-lubricant combination at each aging temperature. The aluminum blocks were heated in large convection ovens for 14 days at their respective temperatures. Oven cavities as well as internal temperatures of the blocks were monitored throughout the aging tests.

EVALUATION OF AGING TESTS

Aged sealed tubes were evaluated by combinations of the following methods:

Visual Inspection

After completion of aging sealed tubes were inspected visually. Recorded were:

- changes in color of the liquid phase scaled per ASTM Standard D1500
- appearance of insoluble reaction products, if any
- appearance of the metal coupons.

In addition, a <u>photographic</u> record was prepared of each test set consisting (from left to right in photograph) of a clean unused steel coupon, the aged coupon, and the sealed tube. The aged coupons for the photographs came from the GC analysis tubes. Photographs were taken as quickly as possible after opening of the tubes to minimize any effects of air oxidation on the appearance of metal surfaces, a precaution especially important for coupons from HCFC tests. Depending on the amount of chloride ion present, many of the HCFC exposed coupons corroded badly upon exposure to air.

Gas Chromatographic (GC) Analysis

All GC analyses were performed on a Gow-Mac Series 550P Thermal Conductivity Detector instrument using a Poropak "Q" column of 2.44 m (8') length. Gas samples were injected through a six-way Valco gas sample valve. Aged sealed tubes to be analyzed were connected to the gas handling manifold with a "tube breaker" assembly. After freezing of the tube contents in liquid nitrogen, the tube tips (scratched with a sharp file) were broken off. The tubes were then warmed slowly and all volatile tube contents were expanded into the manifold, which has a free volume of 1058 ml (64.4 in^3) . A sample of gas from this volume was injected through the sample valve into the GC column. The vapor pressure of the gas sample normally was in the 100 to 200 torr range, depending on the vapor pressure of the test refrigerant.

The column was operated under the following conditions:

Column and injector temperatures: 120°C (R-32 which eluted at 0.55-0.60 min. was also analyzed at 100°C [0.7 min peak] for better resolution of peak(s) below 0.5 min.); Detector temperature: 200°C; Detector current: 150 mA; Carrier gas: Helium; Gas flow rate: 27 ml/min; Sample size: 500 microliter

GC results, i.e. peak retention times, peak areas and peak heights, and area percent for each measurable peak in a chromatogram, were obtained on a Hewlett-Packer Model 3390A Integrator. GC elution peaks of known identities are reported by their chemical designations, while those unknown to us are referred to by their peak retention times.

Chloride and Fluoride Ion Determinations

Chloride and/or fluoride ion contents of sealed tubes were determined with Orion "Ion Selective Electrodes" coupled to a Fisher "Accumet pH Ion Analyzer".

Theory predicts and past experience has shown that the chloride atoms of CFC and HCFC molecules are significantly more reactive than are their fluoride atoms. Therefore, HCFC containing mixtures were analyzed only for chloride ion content in the belief that this is a more sensitive indicator of chemical reactivity. All HFC containing mixtures were of course analyzed for fluoride ion. These determinations were performed in duplicate for every test set. All tube contents, i.e. refrigerant, lubricant, metal coupon, and all glass tube fragments were included in the halide ion determinations.

Results are reported in micrograms of halide ion. From this value we computed "% Refrigerant Reacted" based on the assumption that each halide ion found is the product of the removal of one halide atom from a refrigerant molecule. This assumption is not valid for very reactive systems, where more than one halide atom is removed from refrigerant molecules. R-123 at elevated temperature levels belongs to this latter category.

Oil Analysis

The type of analyses performed varied according to the lubricant tested, as follows:

Hydrocarbon Lubricants. Visual observations are recorded and color photographs were prepared to document the color changes of the aged fluids and the formation of decomposition products such as oil sludges, etc. This is consistent with past practices where sealed tube analysis by chemical methods has generally been confined to the analysis of refrigerants only.

- 16 -

Pentaerythritol Esters. The primary evaluation was the determination of acid numbers following the procedures proscribed in ASTM Method D664 "Test of Neutralization Number by Potentiometric Titration".

Where sufficient sample was available an Infrared (IR) spectrum was prepared. The spectrum of the aged oil was then compared to that of unused lubricant, shown as "Differential Spectra" in the Appendix B figures. All spectra were obtained from neat oil samples on NaCl plates.

Note: The interpretation of the IR spectra is limited to statements about gross spectral changes. A more detailed analysis requires an understanding of ester composition as well as that of potential decomposition products, which is beyond the scope of this project.

Polyalkylene Glycols. Size exclusion chromatography (SEC) was used to assess gross molecular changes of aged PAG lubricants. This technique, even more so than IR analysis, greatly benefits from detailed knowledge of the composition of the base materials and that of their potential decomposition products - again a task beyond the scope of this project.

All chromatograms were obtained using chloroform as eluent. This solvent was selected to avoid interference from residual refrigerant. The columns used were two 300 mm length sets of "Phenogel Linear Mix", with an I.D. of 2 mm. Detection was by refractive index.

COMPLIANCE WITH AGREEMENT

All work performed during this project was in full compliance with the requirements of the original Work Statement or as amended during the course of this project.

The specific changes requested by ARTI were as follows,

(1) the substitution of R-134 for the originally proposed, but unavailable, E-134.

(2) the addition of four sets of sealed tube stability tests involving R-11 and R-123 using the identical test protocol as defined in the work statement, except for the substitution of the $175^{\circ}C$ aging temperature by $160^{\circ}C$.

PRINCIPAL INVESTIGATOR EFFORT

The principal investigator participated in the following Contract activities,

- the procurement and purity analysis of all refrigerant and lubricant samples

- scheduling and supervision of sealed tube preparation
- scheduling and supervision of analysis of aged tubes
- evaluation of test results

- preparation of monthly progress letters, abstracts for presentations at Purdue Compressor Conference and at ASHRAE/ARI Winter Meeting and Show in Chicago in January of 1993, two quarterly progress reports and the final report. APPENDIX A

TABLE A-1: STABILITY OF R-22 WITH MINERAL OIL (150 SSU)

SEALED TUBE CONTENTS:

VALVE STEEL STRIP 1.0 cc (0.9 g) OF TEST LUBRICANT 0.9 g OF R-22

AGING SCHEDULE:

14 DAYS AT TEMPERATURES SHOWN

TEST RESULTS:

	VISUAL APPEARA	NCE OF AGED TUBES	CHEM	ICAL ANAL	<u>YSIS</u>
AGING TEMP, °C <u>(°F)</u>	COLOR CHANGE OF OIL (ASTM)	OTHER VISUAL OBSERVATIONS	GC	CL ION (UG)	%R-22 REACTED
150 (300)	FROM 2.5 To 3.0	STEEL: NO CHANGE	NO CHANGE	2.5	0.001
175 (347)	FROM 2.5 To 3.0	STEEL: NO CHANGE	NO CHANGE	10.0	0.003
200 (392)	FROM 2.5 To 3.0	DARKENING OF STEEL	NO CHANGE	16.7	0.004





RUH B	12	F	EB/26/92	14:56:53					
AREAZ RT	AREA	TYPE	AR/HT 8 852	AREA%	RUN 8	16	Ĥ	PR/15/92	15
1.28	1784898	P 8	0.140	99.865	AREAZ RT	AREA	TYPE	ar/ht	
TOTAL A	REA= 170	7190			9.33 1.25	929 1738299	BB PB	0.051 0.139	

TOTAL AREA= 1731100 NUL FACTOR= 1.0000E+00

TABLE A-2:STABILITY OF R-32 WITH PENTAERYTHRITOL
ESTER (MIXED ACID I)

SEALED TUBE CONTENTS:

VALVE STEEL STRIP 1.0 cc (1.0 g) OF TEST LUBRICANT 1.0 g OF R-32

AGING SCHEDULE:

14 DAYS AT TEMPERATURES SHOWN

TEST RESULTS:

VISUAL APPEARANCE

CHEMICAL ANALYSIS

AGING TEMP. °C, (°F)	OIL COLOR CHANGE (ASTM)	OTHER VISUAL CHANGES	GC	F ION (UG)	%R-32 REACTED	OIL ACID NO.	OIL IR (DIFFERENTIAL SCAN)
150 (300)	NO CHANGE AT 2.5	STEEL: NO CHANGE	NO CHANGE	6.0	0.002	0.00	NO SIGNIFICANT CHANGES
175 (347)	NO CHANGE AT 2.5	STEEL: NO CHANGE	NO CHANGE	4.7	0.001	0.00	NO SIGNIFICANT CHANGES
200 (392)	FROM 2.5 TO 3.0	STEEL: NO CHANGE	NO CHANGE	4.2	0.001	1.67	NO SIGNIFICANT CHANGES

(FLUORIDE ANALYSIS BLANK: 3.1)





TOTAL AREA= 2142800 NUL FACTOR= 1.0000E+00

TABLE A-3:STABILITY OF R-32 WITH POLYPROPYLENE
GLYCOL (BUTYL MONOETHER)

SEALED TUBE CONTENTS:

VALVE STEEL STRIP
1.0 cc (1.0 G) OF TEST LUBRICANT
1.0 G OF R-32

AGING SCHEDULE:

14 DAYS AT TEMPERATURES SHOWN

TEST RESULTS:

VISUAL APPEARANCE

CHEMICAL ANALYSIS

AGING TEMP. °C <u>(°F)</u>	OIL COLOR CHANGE (ASTM)	OTHER VISUAL CHANGES	GC	F ION (UG)	%R-32 REACTEI	SIZE EXCLUSION CHROMATOGRAPHY D OF OIL			
150 (300)	NO CHANGE AT 2.0	STEEL: NO CHANGE	NO CHANGE	0.2	0.000	MINOR CHANGES IN STRUCTURE OF PAG			
175 (347)	NO CHANGE AT 2.5	STEEL: NO CHANGE	NO CHANGE	5.4	0.002	MINOR CHANGES IN STRUCTURE			
200 (392)	NO CHANGE AT 2.5	STEEL: NO CHANGE	NO CHANGE	14.1	0.004	MINOR CHANGES IN STRUCTURE			
(FLUORIDE ANALYSIS BLANK: 3.1)									

```
Tube CC-01 R32/PPG (butyl mono-
                    ether)
         14 Days @ 150°C
```





FIGURE A-06

TABLE A-4: STABILITY OF R-123 WITH MINERAL OIL (ISO VG32)

SEALED TUBE CONTENTS:

VALVE STEEL STRIP 1.0 cc (1.0 g) OF TEST LUBRICANT 0.9 g OF R-123

AGING SCHEDULE:

14 DAYS AT TEMPERATURES SHOWN

TEST RESULTS:

	VISUAL APPEA	RANCE OF TUBES	CHEMICAL ANALYSIS			
AGING TEMP. °C, <u>(°F)</u>	OIL COLOR CHANGE (ASTM)	OTHER VISUAL OBSERVATIONS	GC - % OF R-123 REACTED (UNCORRECTED), PRODUCTS FOUND	CL ION (UG)	%R-123 REACTED (FROM CL ION)	
105 (220)	FROM 2.5 TO 3.0	STEEL: BROWN STAIN	NO CHANGE	98.7	0.045	
	FROM 2. 5 TO 3.0	STEEL: BROWN	NO CHANGE	65.8	0.032	
150 (300)	FROM 2.5 TO 7.0	STEEL: ETCHED	1.8 R-133A	3,260	1.56	
	FROM 2.5 TO 7.5	STEEL: LIGHT SLUDGE DEPOSITS	3.4 R-133A	5,660	2.71	
160 (320)	FROM 2.5 TO >8.0 (BLACK)	BLACK SLUDGE, STEEL: SLUDGE DEPOSITS	14.0 R-133A 0.45 R-143A	54,400	26.0	
175 (347)	FROM 2.5 TO >8.0 (BLACK)	BLACK SLUDGE, STEEL: ETCHED, SLUDGE DEPOSITS	44.5 R-133A 2.0 R-143A	169,440	40.5	

NOTE: SEE PHOTOGRAPHIC RECORD OF VISUAL OBSERVATIONS IN APPENDIX D



101AL AREA4 1300000 NUL FACTUR# 1.000000000







A-13

FIGURE A8.1



A-14
TABLE A-5: STABILITY OF R-124 WITH ALKYLBENZENE OIL

SEALED TUBE CONTENTS:

VALVE STEEL STRIP 1.0 cc (0.9 G) OF TEST LUBRICANT 0.9 G OF R-124

AGING SCHEDULE:

14 DAYS AT TEMPERATURES SHOWN

TEST RESULTS:

	VISUAL APPEARANCE OF TUBES		CHEMICAL ANA	<u>ALYSIS</u>	
AGING TEMP, °C <u>(°F)</u>	OIL COLOR CHANGE (ASTM)	OTHER VISUAL OBSERVATIONS	GC	CL ION (UG)	%R-124 REACTED
150 (300)	NO CHANGE AT 2.0	STEEL: NO CHANGE	NO CHANGE	18.4	0.008
175 (347)	FROM 2.0 TO 2.5	STEEL: NO CHANGE	NO CHANGE	36.1	0.015
200 (392)	FROM 2.0 TO 3.0	STEEL: BROWN STAINS	NO MEASURAE CHANGE	BLE 122.7	0.052





MIR_ FACTUR= 1.0000E+00

TABLE A-6:STABILITY OF R-125 WITH POLYPROPYLENE
GLYCOL (BUTYL MONOETHER)

SEALED TUBE CONTENTS:

VALVE STEEL STRIP 1.0 cc (1.0 g) OF TEST LUBRICANT 1.0 g OF R-125

AGING SCHEDULE:

14 DAYS AT TEMPERATURES SHOWN

TEST RESULTS:

VISUAL APPEARANCE

CHEMICAL ANALYSIS

AGING TEMP. °C <u>(°F)</u>	OIL COLOR CHANGE (ASTM)	OTHER VISUAL CHANGES	GC	F ION (UG)	%R-125 REACTED	SIZE EXCLUSION CHROMATOGRAPHY OF OIL
150 (300)	NO CHANGE AT 2.5	STEEL: NO CHANGE	ELUENT AT 0.62 MIN.	1.6	0.001	CHANGES IN PAG STRUCTURE
175 (347)	NO CHANGE AT 2.5	STEEL: NO CHANGE	ELUENT AT 0.62 MIN.	3.0	0.002	CHANGES IN PAG STRUCTURE
200 (392)	NO CHANGE AT 2.5	STEEL: BLACKENED	NO CHANGE	3.1	0.002	CHANGES IN PAG STRUCTURE

(FLUORIDE ANALYSIS BLANK: 3.6)



MR. FACTUR= 1.0000E100



TOTAL AREA= 1778100 MUL FACTOR= 1.0000E+00

TABLE A-7: STABILITY OF R-125 WITH MODIFIED POLYGLYCOL

SEALED TUBE CONTENTS:

VALVE STEEL STRIP 1.0 cc (1.0 G) OF TEST LUBRICANT 1.0 G OF R-125

AGING SCHEDULE:

14 DAYS AT TEMPERATURES SHOWN

TEST RESULTS:

VISUAL APPEARANCE

CHEMICAL ANALYSIS

AGING TEMP. °C, <u>(°F)</u>	OIL COLOR CHANGE (ASTM)	OTHER VISUAL CHANGES	GC	F ION (UG)	%R-125 REACTED	SIZE EXCLUSION CHROMATOGRAPHY OF OIL
150 (300)	FROM 2.5 TO 3.0	STEEL: NO CHANGE	NO CHANGE	11.0	0.007	CHANGES IN PAG STRUCTURE
175 (347)	FROM 2.5 TO 3.5	STEEL: NO CHANGE	NO CHANGE	7.1	0.004	CHANGES IN PAG STRUCTURE
200 (392)	FROM 2.5 TO 4.0	STEED.: NO CHANGE	NO CHANGE	8.5	0.005	CHANGES IN PAG STRUCTURE

(FLUORIDE ANALYSIS BLANK: 8.2)



A-22

Gas Chromatograms R-125/Modified Polyglycol



TABLE A-8:STABILITY OF R-125 WITH PENTAERYTHRITOL
ESTER (MIXED ACID I)

SEALED TUBE CONTENTS:

VALVE STEEL STRIP 1.0 cc (1.0 g) OF TEST LUBRICANT 1.0 g OF R-125

AGING SCHEDULE:

14 DAYS AT TEMPERATURES SHOWN

TEST RESULTS:

VISUAL APPEARANCE

CHEMICAL ANALYSIS

AGING TEMP. °C, <u>(°F)</u>	OIL COLOR CHANGE (ASTM)	OTHER VISUAL CHANGES	GC	F ION (UG)	%R-125 REACTED	oil Acid No.	OIL IR (DIFFERENTIAL SCAN)
150 (300)	FROM 2.5 TO 3.0	STEEL: NO CHANGE	NO CHANGE	8.2	0.005	0.00	NO SIGNIFICANT CHANGES
175 (347)	FROM 2.5 TO 3.0	STEEL: NO CHANGE	NO CHANGE	7.7	0.005	0.48	NO SIGNIFICANT CHANGES
200 (392)	FROM 2.5 TO 3.0	STEEL: NO CHANGE	NO CHANGE	7.7	0.005	1.09	CHANGES IN 12001300 CM ⁻¹ REGION

(FLUORIDE ANALYSIS BLANK: 3.1)





HUL FACIOR= 1.0009E+00

TABLE A-9:STABILITY OF R-134A WITH POLYPROPYLENE
GLYCOL (BUTYL MONOETHER)

SEALED TUBE CONTENTS:

VALVE STEEL STRIP 1.0 cc (1.0 G) OF TEST LUBRICANT 1.0 G OF R-134A

AGING SCHEDULE:

14 DAYS AT TEMPERATURES SHOWN

TEST RESULTS:

VISUAL APPEARANCE CHEMICAL ANALYSIS AGING TEMP SIZE EXCLUSION **OIL COLOR** OTHER FION %R-134A °C **CHROMATOGRAPHY** CHANGE VISUAL (°F) (ASTM) CHANGES GC (UG) REACTED OF OIL 150 NO CHANGE STEEL: NO NO CHANGE 1.2 0.001 CHANGES IN PAG AT 2.5 (300)CHANGE STRUCTURE 175 NO CHANGE STEEL: NO NO CHANGE 1.2 0.001 **CHANGES IN PAG** (347) AT 2.5 CHANGE STRUCTURE 200 **FROM 2.5** STEEL: NO CHANGE 1.1 0.001 CHANGES IN PAG (392) TO 3.0 DARKENED STRUCTURE

(FLUORIDE ANALYSIS BLANK: 3.1)



510r

eini i	11	Н	MI/16/92	14:02:49
npt ny R t n 34 1, nz	nren 1812 2890280	tyre 89 78	(R/II) 9.952 9.149	AREA2 0.465 99.935

10101_0KF0= E892009 MN_1_0CT0R= 1.0000E100



TABLE A-10:STABILITY OF R-134A WITH POLYPROPYLENE
GLYCOL DIOL

SEALED TUBE CONTENTS:

VALVE STEEL STRIP 1.0 cc (1.0 G) OF TEST LUBRICANT 1.0 G OF R-134A

AGING SCHEDULE:

14 DAYS AT TEMPERATURES SHOWN

TEST RESULTS:

VISUAL APPEARANCE

CHEMICAL ANALYSIS

AGING TEMP. °C, <u>(°F)</u>	OIL COLOR CHANGE (ASTM)	other Visual Changes	GC	F ION (UG)	%R-134A REACTED	SIZE EXCLUSION CHROMATOGRAPHY OF OIL
150 (300)	NO CHANGE AT 2.0	STEEL: NO CHANGE	NO CHANGE	0.6	0.000	SIGNIFICANT CHANGES IN PAG STRUCTURE
175 (347)	NO CHANGE AT 2.0	STEEL: NO CHANGE	NO CHANGE	2.1	0.001	SIGNIFICANT CHANGES IN PAG STRUCTURE
200 (392)	FROM 2.0 TO 2.5	STEEL: NO CHANGE	NO CHANGE	0.1	0.000	SIGNIFICANT CHANGES IN PAG STRUCTURE

(FLUORIDE ANALYSIS BLANK: 8.1)

FIN R-134n "An Received" ļ (NANP_1.110)_ J1.# Tube CJ-01 R-134a/Polypropylene-glycol Diol 14 Days @ 150°C STARI . 36 89.1 C 1.14 **STOP** 13:38 4 RUH 8 JUN/24/92 13:35:10 AREA% AR/HT AREAZ AREA TYPE _R1 9.0.76 99.924 FB PB 0.055 0.121 8.36 1456 1911**700** 1.08 101AL AREA= 1913200 NUL FACTOR= 1.0000E+00 . 5111

rint a st			J	W/16/92	14:02:49		
npt n : Nt N		nht n 1912 Aunuero	Int Pa	1 .73			

A-31





Gas Chromatograms R-134a/Polypropyleneglycol Diol

10101 0EF0= 1916489 MIN FOCTOR= 1 88886189

TABLE A-11: STABILITY OF R-134A WITH MODIFIED POLYGLYCOL

SEALED TUBE CONTENTS:

VALVE STEEL STRIP 1.0 cc (1.0 G) OF TEST LUBRICANT 1.0 G OF R-134A

AGING SCHEDULE:

14 DAYS AT TEMPERATURES SHOWN

TEST RESULTS:

VISUAL APPEARANCE

CHEMICAL ANALYSIS

AGING TEMP. °C, <u>(°F)</u>	OIL COLOR CHANGE (ASTM)	OTHER VISUAL CHANGES	GC	F ION (UG)	%R-134A REACTED	SIZE EXCLUSION CHROMATOGRAPHY OF OIL
150 (300)	FROM 2.5 TO 3.0	STEEL: NO CHANGE	ELUENT AT 0.08% AT 1.0 MIN.	1.1	0.001	SIGNIFICANT CHANGES IN PAG STRUCTURE
175 (347)	FROM 2.5 TO 3.5	STEEL: NO CHANGE	0.05% AT 1.0 MIN.	2.3	0.001	SIGNIFICANT CHANGES IN PAG STRUCTURE
200 (392)	FROM 2.5 TO 4.0	STEEL: NO CHANGE	TRACE OF ELUENT AT 1.03 MIN.	2.2	0.001	SIGNIFICANT CHANGES IN PAG STRUCTURE

(FLUORIDE ANALYSIS BLANK: 8.2)





AREA%				
RT	AREA	TYPE	nr/ht	AREA%
0.40	983	FB	0.057	0.019
1.03	1070	BP	0.103	0.053
1.29	2011300	PB	0.143	99.8 98

TOTAL AREA= 2013400 MIL FACTOR= 1.0000E+00

TABLE A-12:STABILITY OF R-134A WITH PENTAERYTHRITOL
ESTER (MIXED ACID II)

SEALED TUBE CONTENTS

VALVE STEEL STRIP 1.0 cc (1.0 g) OF TEST LUBRICANT 1.0 g OF R-134A

AGING SCHEDULE:

14 DAYS AT TEMPERATURES SHOWN

VISUAL APPEARANCE

TEST RESULTS:

AGING TEMP. °C, <u>(°F)</u>	OIL COLOR CHANGE (ASTM)	OTHER VISUAL CHANGES	GC	F ION (UG)	%R-134A REACTED	OIL ACID NO.	OIL IR (DIFFERENTIAL SCAN)
150 (300)	NO CHANGE AT 2.5	STEEL: NO CHANGE	NO CHANGE	0.8	0.000	0.00	NO CHANGE
175 (347)	NO CHANGE AT 2.5	STEEL: LIGHT STAIN	NO CHANGE	0.8	0.000	0.00	NO CHANGE
200 (392)	FROM 2.5 TO 3.0	STEEL: DARKENED	NO CHANGE	0.7	0.000	0.67	NO CHANGE

CHEMICAL ANALYSIS

(FLUORIDE ANALYSIS BLANK: 3.1)





MUL FACTUR= 1.0000E+00

TABLE A-13:STABILITY OF R-134A WITH PENTAERYTHRITOL
ESTER (MIXED ACID I)

SEALED TUBE CONTENTS:

VALVE STEEL STRIP 1.0 cc (1.0 g) OF TEST LUBRICANT 1.0 g OF R-134A

AGING SCHEDULE:

14 DAYS AT TEMPERATURES SHOWN

TEST RESULTS:

VISUAL APPEARANCE

CHEMICAL ANALYSIS

AGING TEMP. °C, <u>(°F)</u>	OIL COLOR CHANGE (ASTM)	OTHER VISUAL CHANGES	GC	F ION (UG)	%R-134A REACTED	OIL ACID NO.	OIL IR (DIFFERENTIAL SCAN)
150 (300)	NO CHANGE AT 2.5	STEEL: NO CHANGE	NO Change	2.5	0.001	0.00	NO CHANGE
175 (347)	NO CHANGE AT 2.5	STEEL: NO CHANGE	no Change	1.4	0.001	0.00	NO CHANGE
200 (392)	FROM 2.5 TO 3.0	STEEL: NO CHANGE	NO CHANGE	1.8	0.001	0.74	NO CHANGE

(FLUORIDE ANALYSIS BLANK: 3.1)







STOP

RUN	14	W	k/04/9 2	12:07:46
AREAZ RT 0.34 1.10	AREA 1221 1853200	TYPE BB PB	AR/HT 0.054 0.128	AREAZ 0.066 99.934

TABLE A-14:STABILITY OF R-134A WITH PENTAERYTHRITOL
ESTER (100 cSt)

SEALED TUBE CONTENTS:

VALVE STEEL STRIP 1.0 cc (1.0 g) OF TEST LUBRICANT 1.0 g OF R-134A

AGING SCHEDULE:

14 DAYS AT TEMPERATURES SHOWN

TEST RESULTS:

VISUAL APPEARANCE

CHEMICAL ANALYSIS

AGING TEMP. °C <u>(°F)</u>	OIL COLOR CHANGE (ASTM)	OTHER VISUAL CHANGES	GC	F ION (UG)	%R-134A REACTED	oil Acid No.	OIL IR (DIFFERENTIAL SCAN)
150 (300)	NO CHANGE AT 3.0	STEEL: NO CHANGE	NO CHANGE	1.6	0.001	0.94	SPLIT C=O AT 1700 To 1750 CM ^{.1}
175 (347)	FROM 3.0 TO 3.5	STEEL: DARKENED	NO CHANGE	1.0	0.001	0.63	SPLIT C=O 1700 To 1750 ^{CM-1}
200 (392)	FROM 3.0 TO 4.5	STEEL: BLACKENED	0.56% OF CO ₂ FORMED	6.8	0.004	0.99	AS ABOVE, PLUS C-O ABSORPTION AROUND 1200 CM-1

COMMENTS ON IR ANALYSIS: SPLIT CARBONYL BANDS INDICATE FORMATION OF EITHER NEW ACID MOLECULES (TRANSESTERIFICATION), OR ALDEHYDES AND KETONES, I.E. DESTRUCTION OF ESTER LINKAGES. AT 200 C ALCOHOLS ARE FORMED (C-O ABSORPTION) TOGETHER WITH THE APPEARANCE OF CO_2 (FROM GC ANALYSIS). THIS INDICATES THE BREAKAGE OF ESTER LINKAGES BY A DECARBOXYLATION MECHANISM.

A - 42



P171 8 11		•	J	W/16/92	14-02-49	
nRF n2 RT 9 , 34 1 , 92		nren 1812 2000200	1 YPE 90 98	nn/nt 9.978 9.149	MtA: 9.065 99.935	

11110L MAFA- 2092009 MR FACINE 1.0000E100



TUTAL AREA= 1823588 NUL FACTOR= 1.8888E+88

TABLE A-15: STABILITY OF R-142B WITH ALKYLBENZENE OIL

SEALED TUBE CONTENTS:

VALVE STEEL STRIP 1.0 CC (0.9 G) OF TEST LUBRICANT 0.9 G OF R-142B

AGING SCHEDULE:

14 DAYS AT TEMPERATURES SHOWN

TEST RESULTS:

VISUAL APPEARA		ARANCE OF TUBES	<u>CHEMIC</u>	CHEMICAL ANALYSIS		
AGING TEMP, °C <u>(°F)</u>	OIL COLOR CHANGE (ASTM)	OTHER VISUAL OBSERVATIONS	GC	CL ION (UG)	%R-142B REACTED	
150 (300)	NO CHANGE AT 2.0	STEEL: NO CHANGE	NO CHANGE	34.8	0.011	
175 (347)	FROM 2.0 TO 2.5	STEEL: SLIGHT DARKENING	NO CHANGE	45.0	0.014	
200 (392)	FROM 2.0 TO 3.0	STEEL: ETCHED AND STAINED	NO MEASURABLE CHANGE	525	0.169	



TOTAL AREA= 2520400 HUL FACTOR= 1.0000E+00

101AL AREA= 1031600 WUL FAC10R= 1.0000E+00





TABLE A-16:STABILITY of R-143A WITH PENTAERYTHRITOL
ESTER (MIXED ACID I)

SEALED TUBE CONTENTS:

VALVE STEEL STRIP 1.0 cc (1.0 G) OF TEST LUBRICANT 1.0 G OF R-143A

AGING SCHEDULE:

14 DAYS AT TEMPERATURES SHOWN

VISUAL APPEARANCE

TEST RESULTS:

AGING TEMP. °C, <u>(°F)</u>	OIL COLOR CHANGE (ASTM)	OTHER VISUAL CHANGES	GC	F ION (UG)	%R-143A REACTED	OIL ACID NO.	OIL IR (DIFFEREN- TIAL SCAN)
150 (300)	FROM 2.5 TO 3.0	STEEL: NO CHANGE	NO CHANGE	2.6	0.001	0.51	NO CHANGE
175 (347)	FROM 2.5 TO 3.0	STEEL: NO CHANGE	NO CHANGE	4.1	0.002	1.07	NO CHANGE
200 (392)	FROM 2.5 TO 3.0	STEEL: BLACKENED	NO CHANGE	1.1	0.001	0.99	NO CHANGE

CHEMICAL ANALYSIS

(FLUORIDE ANALYSIS BLANK: 3.6)





KUN T	15	JUN/11/92		
AREAZ RT 0.36 0.51 0.86	AREA 4126 1924 2182888	TYPE PV VB PB	AR/HT 9.053 9.053 9.101	AREA 8.189 8.847 99.765

TOTAL AREA= 2187100 NUL FACTOR= 1.0000E+00
TABLE A-17: STABILITY OF R-152A WITH ALKYLBENZENE

SEALED TUBE CONTENTS:

VALVE STEEL STRIP 1.0 cc (0.9 G) OF TEST LUBRICANT 0.9 G OF R-152A

AGING SCHEDULE:

14 DAYS AT TEMPERATURES SHOWN

TEST RESULTS:

AGING TEMP, °C _(°F)	COLOR CHANGE OF OIL (ASTM)	OTHER VISUAL OBSERVATIONS	GC	F ION (UG)	%R-152A REACTED
150 (300)	FROM 2.0 TO 2.5	STEEL: NO CHANGE	NO CHANGE	0.4	0.000
175 (347)	FROM 2.0 TO 2.5	STEEL: NO CHANGE	NO CHANGE	2.0	0.001
200 (392)	FROM 2.0 TO 3.0	STEEL: DARKENED	NO CHANGE	1.1	0.000
		(FLUORIDE ANALYSIS BLANK:		3.1)	

VISUAL APPEARANCE OF TUBES CHEMICAL ANALYSIS

NOTE: R-152A AND ALKYLBENZENE LUBRICANT WERE NOT MISCIBLE AT ROOM TEMPERATURE. ALL TUBES EXHIBITED TWO DISTINCT LIQUID PHASES.

R-152a "As Received"

.

14 Days @ 150°C



TOTAL AREA= 2147500 NUL FACTOR= 1.0000E+00





TABLE A-18:STABILITY OF R-134 WITH PENTAERYTHRITOL
ESTER (MIXED ACID I)

SEALED TUBE CONTENTS:

VALVE STEEL STRIP 1.0 cc (1.0 g) OF TEST LUBRICANT 1.0 g OF R-134

AGING SCHEDULE:

14 DAYS AT TEMPERATURES SHOWN

TEST RESULTS:

VISUAL APPEARANCE

CHEMICAL ANALYSIS

AGING TEMP. °C, <u>(°F)</u>	OIL COLOR CHANGE (ASTM)	OTHER VISUAL CHANGES	GC	F ION (UG)	%R-134 REACTED	OIL ACID No.	OIL IR (DIFFEREN- TIAL SCAN)
150 (300)	NO CHANGE AT 2.5	STEEL: NO CHANGE	NO CHANGE	1.2	0.001	0.59	NO SIGNIF. CHANGES
175 (347)	FROM 2.5 TO 3.0	STEEL: NO CHANGE	NO CHANGE	0.1	0.000	0.83	NO SIGNIF. CHANGES
200 (392)	FROM 2.5 TO 3.0	STEEL: DARKENED	NO CHANGE	0.7	0.000	1.69	NO SIGNIF. CHANGES

(FLUORIDE ANALYSIS BLANK: 3.6)



PB PB	9.056 9.133	нкенz 0.035 99.965	r 11 11
			Π

•	TOTAL AREA=	1868800
	NUL FACTOR=	1.0000E+00

TOTAL AREA= 1984998 MUL FACTOR= 1.88888488

0.36 1.19 666 1984298





A-56

TABLE A-19: STABILITY OF R-11 WITH MINERAL OIL (ISO VG32)

SEALED TUBE CONTENTS:

VALVE STEEL STRIP 1.0 cc (1.0 G) OF TEST LUBRICANT 0.9 G OF R-11

AGING SCHEDULE:

14 DAYS AT TEMPERATURES SHOWN

TEST RESULTS:

	VISUAL APPEARANCE OF TUBES		<u>CHEMICAL</u>		
AGING TEMP. °C <u>(°F)</u>	OIL COLOR CHANGE (ASTM)	OTHER VISUAL OBSERVATIONS	GC -% OF R-11 REACTED (UNCORRECTED), PRODUCTS FOUND	CL ION (UG)	%R-11 REACTED (FROM CL ION)
105 (220)	FROM 2.5 TO 4.5	STEEL: LIGHT STAINS	NO MEASURABLE CHANGES	1,063	0.46
150 (300)	FROM 2.5 TO >8.0 (BLACK)	OIL: GELLED, STEEL: COVERED WITH SLUDGE	72.0 R-21, <0.5 R-31, 1.0 CO ₂	224,600	72.0
180 (320)	FROM 2.5 TO >8.0 (BLACK)	OIL: GELLED, STEEL: COVERED WITH SLUDGE	R-11 HAS DISAPPEARED, 47.8 R-21, 17.2 R-31, 31.4 CO ₂	270,200	87.0

NOTE: SEE PHOTOGRAPHIC RECORD OF VISUAL OBSERVATIONS IN APPENDIX D

A - 57



101AL AREA= 1940500 NUL FACTOR= 1.0000E+00

•

FIGURE A-38



TABLE A-20:STABILITY OF R-11 WITH WHITE NAPHTHENIC
MINERAL OIL (ISO VG46)

SEALED TUBE CONTENTS:

VALVE STEEL STRIP 1.0 cc (1.0 G) OF TEST LUBRICANT 0.9 G OF R-11

AGING SCHEDULE:

14 DAYS AT TEMPERATURES SHOWN

TEST RESULTS:

	VISUAL APPEARANCE OF TUBES		CHEMICAL ANALYSIS			
AGING TEMP. °C, <u>(°F)</u>	OIL COLOR CHANGE (ASTM)	OTHER VISUAL OBSERVATIONS	GC-% OF R-11 REACTED (UNCORRECTED), PRODUCTS FOUND	CL ION (UG)	%R-11 REACTED (FROM CL ION)	
105 (220)	NO CHANGE FROM 2.0	STEEL: LIGHT STAINS	NO MEASURABLE CHANGE	33	0.014	
150 (300)	FROM 2.0 TO >8.0 (BLACK)	STEEL: ETCHED, LIGHT SLUDGE	17.8 R-21 1.37 R-31 CO ₂	94,500	45.2	
160 (320)	FROM 2. 0 TO >8.0 (BLACK)	STEEL: ETCHED, OIL: GELLED	R-11: ALMOST DISAPPEARED, 75.0 R-21 10.4 R-31 13.7 CO ₂	202,500	97.0	

NOTE: SEE PHOTOGRAPHIC RECORD OF VISUAL OBSERVATIONS IN APPENDIX D

FIGURE A-39





TABLE A-21: STABILITY OF R-123 WITH WHITE NAPHTHENIC MINERAL OIL (ISO VG46)

SEALED TUBE CONTENTS:

VALVE STEEL STRIP 1.0 cc (1.0 G) OF TEST LUBRICANT 0.9 G OF R-123

AGING SCHEDULE:

14 DAYS AT TEMPERATURES SHOWN

TEST RESULTS:

	VISUAL APPEARANCE OF TUBES		CHEMICAL ANALYSIS			
AGING TEMP. °C <u>(°F)</u>	OIL COLOR CHANGE (ASTM)	OTHER VISUAL OBSERVATIONS	GC-% OF R-123 REACTED (UNCORRECTED), PRODUCTS FOUND	CL ION (UG)	%R-123 REACTED (FROM <u>CL ION)</u>	
105 (220)	NO CHANGE FROM 2.0	STEEL: NO CHANGE	NO CHANGE	48	0.002	
150 (300)	FROM 2.0 TO 7.5	STEEL: LIGHT DEPOSITS (OIL SLUDGE)	3.9 R-133A 0.08 R-143A	12,200	5.84	
160 (320)	FROM 2.5 TO >8.0 (BLACK)	STEEL: ETCHED, SLUDGE DEPOSITS	10.9 R-133A 0.38 R-133A CO ₂	37,000	8.9	

NOTE: SEE PHOTOGRAPHIC RECORD OF VISUAL OBSERVATIONS IN APPENDIX D

FIGURE A-41





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APPENDIX B

IR Spectra of Pentaerythritol Ester (mixed acid I)aged with R-32

- - - Unused Oil - - -

- - Oil aged for 14 Days at 200°C with R-32

- - Differential Scan: Aged Oil vs. Unused Oil





Size Exclusion Chromatogram: PPG butyl monoether/R-32, 14 days @ 150°C

R-17/FFG (hutyl monnether) Aged 011 150°C (Tube #CC-02)



в – З







в – 5



в – б

IR Spectra Pentaerythritol Ester (mixed acid I) after Aging at 150°C with R-125.

---Unused PE ester, acids

--- Aged PE ester, acids

--- Differential Scan: Aged Oil vs. New Oil

в – 7



IR Spectra of Pentaerythritol Ester after Aging at 200°C with R-125.

--- Unused PE ester, mixed acids I











B - 10



FIGURE B-11

Size Exclusion Chromatogram: PPG Diol/R-134a, 14 days at 175°C

Polypropyleneglycol Diol - Aged Oil, 175°C (Tube CJ-12)

Size Exclusion Chromatograms: PPG Diol/R-134a, 14 days at 200°C





B - 12



Hodified Polyglycol - Aged Oil, 150°C (Tube CK-02)

B - 13



Size Exclusion Chromatograms:

Modified Polyglycol/R-134a, 14 days at 175°C



Hodified Polyglycol - Aged Oil, 150°C (Tube CK-02)

B - 14

Size Exclusion Chromatograms: M

Modified Polyglycol/R-134a, 14 days at 200°C





B - 15



IR Spectra: Pentaerythritol Ester -(mixed acid I)/R-134a 14 days at 200°C

--- Unused Oil

--- Aged Oil, 14 days at 200°C

--- Differential Scan: Aged Oil vs. Unused Oil



B - 17

IR Spectra:

Pentaerythritol Ester (100 cSt)/R-134a, 14 days at 150°C



19

B - 18



• 5

в – 19

IR Spectra: Differential Scans Pentaerythritol Ester (100 cSt)/R-134a - Aged Oils (150°C, 175°C, 200°C) vs. Unused Oil

- - - 150°C



- - - 175°C

- - - 200°C
FIGURE B-21

IR Spectra: Pentaerythritol Ester (mixed acid I)/R-143a 14 days at 200°C

- - Unused Oil

MICRONS MAVELENGTH 100 80 60 40 20 0 1200 CM ' 1000 900 800 700 2000 1500 00 3000 SMECTRUM REMARKS_ LEGEND' ORIGIN_ARTE SPECTRUM NO._ SAMPLE CARLINE MIZ Neference OIL y PURITY_ Z PHASE File on Mcl DATE ___ 42/92 THICKNESS OPERATOR WAYELENGTH IMICRONS) 11 100 80 ٥٥ 40 ΙŢ 20 Jo. 4000 3000 1200 1000 900 800 700 1500 2000 , ru REMARKS LEGEND SPECTRUM NO. SAMPLE SP 22 OIL PURITY_ PHASE File m mcl 4.41/11 DATE OPERATOR _BAR THICKNESS WAYELENGTH MICRONSI 100 80 ž ٨c 60 40 10 20 4000 3000 1200 CM' 2000 1500 1000 900 800 700 SPECTRUM NO. ____ ORIGIN_ATT LEGEND REMARKS Y _ mainelly in estress ş Differential Sean PURITY... eefeen. Z PHASE _ PILE on mict DATE 1/11/11

OPERATOR ____

IHICKNESS

- - Aged Oil, 14 days at 200°C

- - Differential Scan: Aged Oil vs. Unused Oil

B - 21

FIGURE B-22

IR Spectra: Pentaerythritol
Ester- (mixed acid I)/R-134,
14 days at 200°C

- - Unused Oil



THICKNESS

OPERATOR

.....

- - Aged Oil, 14 days at 200°C

- - Differential Scan: Aged vs. Unused Oil APPENDIX C

APPENDIX C LISTS ALL TEST REFRIGERANTS AND LUBRICANTS USED IN THE COURSE OF THIS PROJECT. GENERIC DESIGNATIONS AND THEIR COMMERCIAL EQUIVALENTS ARE CROSS REFERENCED, WHERE APPLICABLE. ALSO DOCUMENTED IN APPENDIX C IS ALL INFORMATION DEVELOPED ON THE PURITY LEVEL OF THE MATERIALS USED.

TABLE C-1: TEST REFRIGERANTS

THE PURITY OF ALL REFRIGERANTS USED WAS DETERMINED BY GC ANALYSIS OF LIQUID PHASE REFRIGERANT SAMPLES. ABSENCE OF ANY MEASURABLE QUANTITIES OF CONTAMINANT GASES, EXCEPT FOR THE SMALL AMOUNTS OF AIR FROM INADVERTENT LEAKS IN THE GAS HANDLING SYSTEM IS INDICATED IN THE TABLE BY THE NOTATION "PURE MATERIAL".

REFRIGERANT	GC RESULTS	REFERENCE CHROMATOGRAM
R-11	PURE MATERIAL	FIG. C-O1
R-22	PURE MATERIAL	FIG. C-02
R-32	PURE MATERIAL	FIG. C-03
R-123	PURE MATERIAL	FIG. C-04
R-124	PURE MATERIAL	FIG. C-05
R-125	UNIDENTIFIED IMPURITY AT <0.5%	FIG. C-06
R-134A	PURE MATERIAL	FIG. C-07
R-134	POSSIBLE IMPURITY ELUTING AFTER R-134	FIG. C-08
R-142B	PURE MATERIAL	FIG. C-09
R-143A	CO ₂ (?) IMPURITY	FIG. C-10
R-152A	PURE MATERIAL	FIG. C-11

TABLE C-2: TEST LUBRICANTS

GENERIC	COMMERCIAL	COMMERCIAL	TOTAL ACID NO	WATER CONTENT	METAL CONTENT (PPM)	
IDENTIFICATION	SOURCE	DESIGNATION	(MG KOH/G)	(PPM)	CU	FE
MINERAL OIL (ISO VG32) - NAPHTHENIC	WITCO CHEMICAL	SUNISO 3GS	0.001	18	1	0
MINERAL OIL (ISO VG46) - NAPHTHENIC WHITE OIL	WITCO CHEMICAL	FREEZENE- NAPHTHENIC HEAVY	0.11	15	2	<1
ALKYLBENZENE	Shrieve Chemical	ZEROL 150	<0.001	17	1	0
POLYPROPYLENE GLYCOL BUTYL- MONOETHER	ICI	EMKAROX (ISO VG32	0.05	23#	<1	<1
Modified Poly- Glycol (propriet. Composition)	ALLIED- SIGNAL	BRL-150	0.13	21#	<1	<1
POLYPROPYLENE GLYCOL DIOL	DOW CHEMICAL	P 425 (ISO VG22)	0.07	16#	<1	<1
PENTAERYTHRITOL ESTER (MIXED ACID I)	CASTROL	ICEMATIC SW32 (ISO VG32)	0.02	24#	<1	<1
PENTAERYTHRITOL ESTER (MIXED ACID II)	ICI	EMKARATE RLE (ISO VG22)	0.08	20#	<1	<1
PENTAERYTHRITOL ESTER (100 cSt)	HENKEL- EMERY	EMERY 4078X (2928 ISO VG 100)	0.10	18#	<1	<1

= AFTER VACUUM DEHYDRATION

Gas Chromatogram of R-11 as Received

R-11 "As Received"



RUN 8	19		301/24/92	17:10:40
NREA2 R1 0.37 2.85		AREA 1914 919 81 1698808 91	M./Ht 8.957 8.960	AREAX 0.053 99.947

1010L AREA= 1699990 MR_ FACTOR= 1.0000E100

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FIGURE C-02





Gas Chromatogram of R-32 as received

R-32 "As Received"



Gas Chromatogram of R-123 as received



•...



INIAL AREA- 3833090 MR. FACTOR+ 1.00000100

C - 4

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FIGURE C-06

Gas Chromatogram of

R-125 as Received

Gas Chromatogram of R-124 as received



Gas Chromatogram of R-134a as Received



FIGURE C-08

Gas Chromatogram of R-134 as Received

С – б

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Gas Chromatogram of R-142b as Received



Gas Chromatogram of R-143a as Received



R-143a "As Received"

*



Note: 0.49 min peak is probably CO₂.

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C - 7

Gas Chromatogram of R-152a as Received

R-152a "As Received"



C - 8

APPENDIX D

TABLE OF CONTENTS

APPENDIX D CONTAINS COPIES OF PHOTOGRAPHS FROM THOSE SEALED TUBE SETS THAT EXHIBITED SIGNIFICANT VISUAL CHANGES AFTER THE THERMAL AGING OF THE RESPECTIVE REFRIGERANT-OIL MIXTURES, AS FOLLOWS:

D-1 R-123/MINERAL OIL	FIGURE D-1
D-2 R-124/ALKYLBENZENE	FIGURE D-2
R-125/MODIFIED POLYGLYCOL	FIGURE D-3
R-134A/MODIFIED POLYGLYCOL	FIGURE D-4
D-5 R-134A/PENTAERYTHRITOL ESTER (100 cSt)	FIGURE D-5
R-11/MINERAL OIL	FIGURE D-6
D-7 R-11/HEAVY NAPHTHENIC	FIGURE D-7
R-123/HEAVY NAPHTHENIC	FIGURE D-8











































